



# CINTECX'S ANNUAL CHALLENGE 2023



## DESIGN AND SYNTHESIS OF NEW MATERIALS FOR H<sub>2</sub> GENERATION AS A RENEWABLE ENERGY SOURCE



### 1. ACTUAL SCENARIO

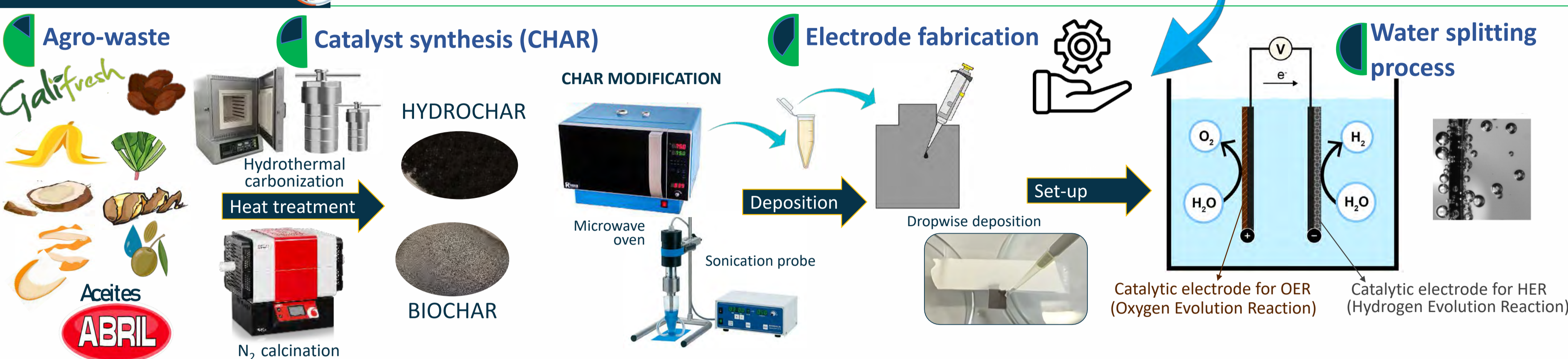


### 2. NOVEL ALTERNATIVE

#### Ecofriendly H<sub>2</sub> production by water splitting process

- Request catalyst is synthetised from agrowastes
  - Favouring the "green" use of water splitting processes
  - Avoiding metal catalysts
- Environmental friendliness  
Enhanced stability  
Cost reduction

### 3. PROCEDURE

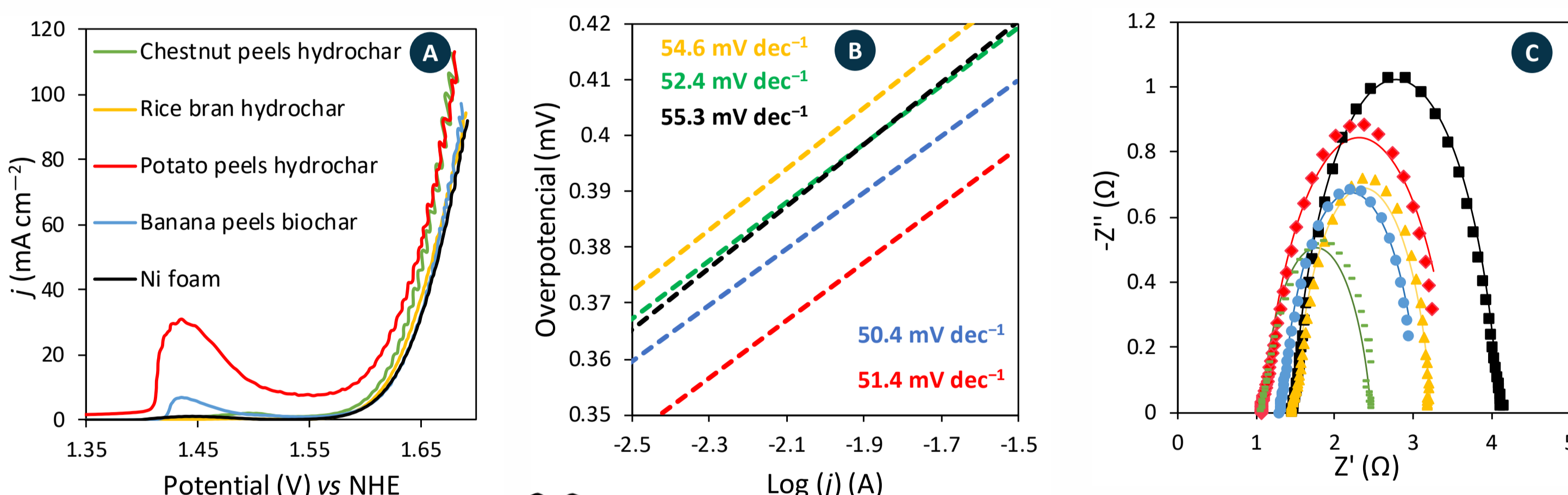


### 4. RESULTS

#### 4.1 CHARS' SCREENING

Banana, potato and chestnut peels, as well as rice bran, spinach stem and spent coffee were thermally-treated. Only some chars exhibited catalytic water splitting behaviour, specifically hydrochars, due to their acid pH and their higher content on oxygenated groups.

Hydrochar synthesized from potato peels and biochar attained from N<sub>2</sub> calcination of banana peels provided higher intensities at a given potential **A** with the smaller activation energy **B**. Concerning electrochemical impedance spectroscopy **C** spinach stem and rice bran hydrochars were also good electric conductors.



#### CHAR CHARACTERISTICS

Metal free (C/N/O/H)  
Functional groups

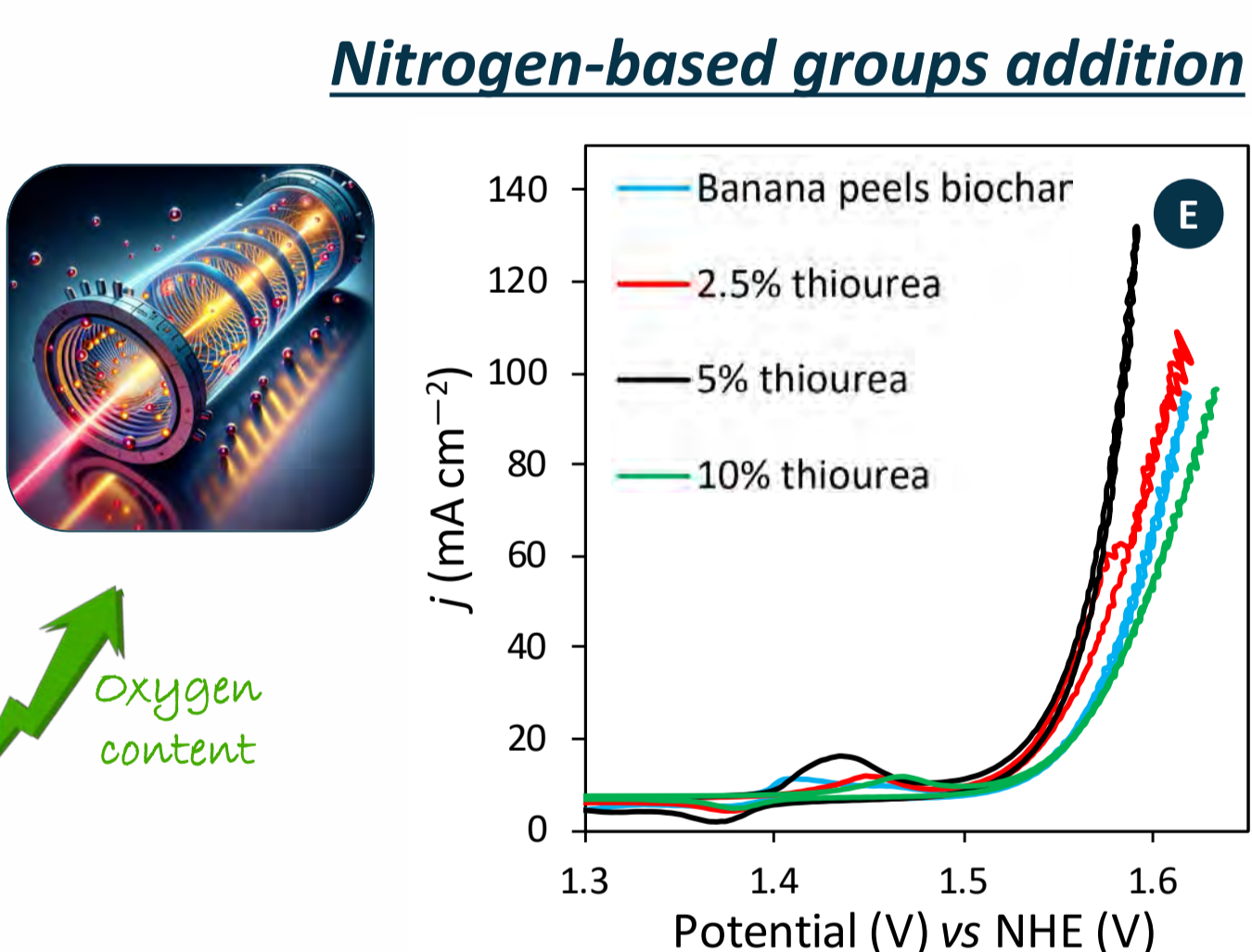
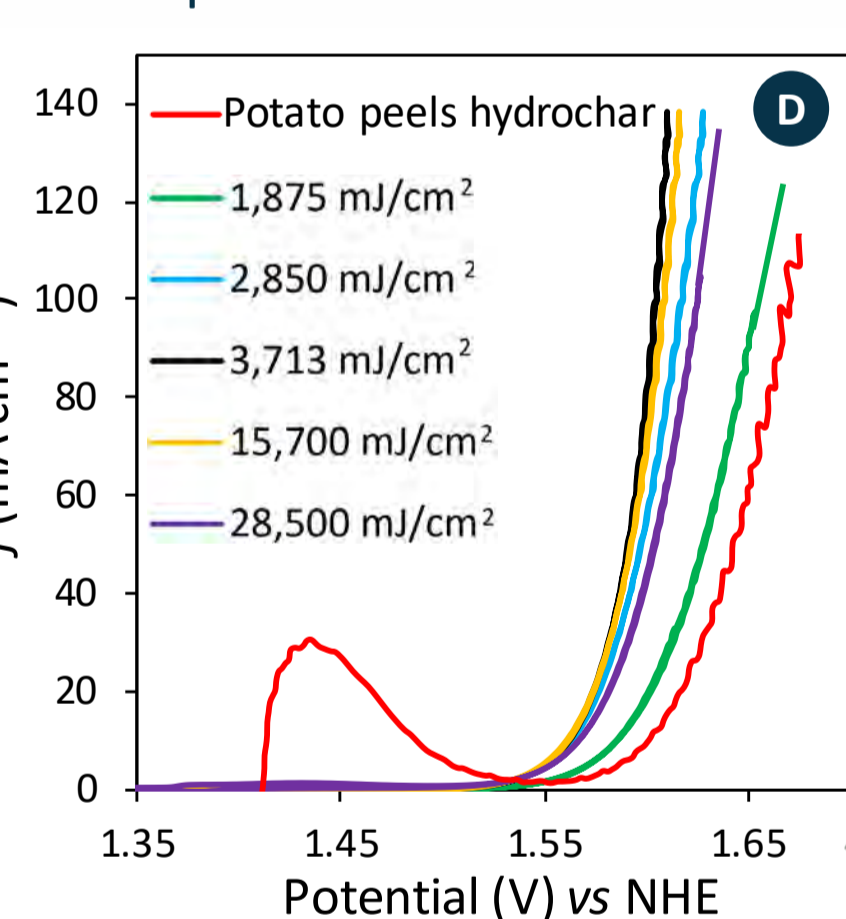
**HYDROCHAR**  
Oxygenated groups  
Amorphous  
Acid pH  
Cavity-shape

**BIOCHAR**  
Nitrogenated groups  
High surface areas  
Basic pH  
Sharp-shape

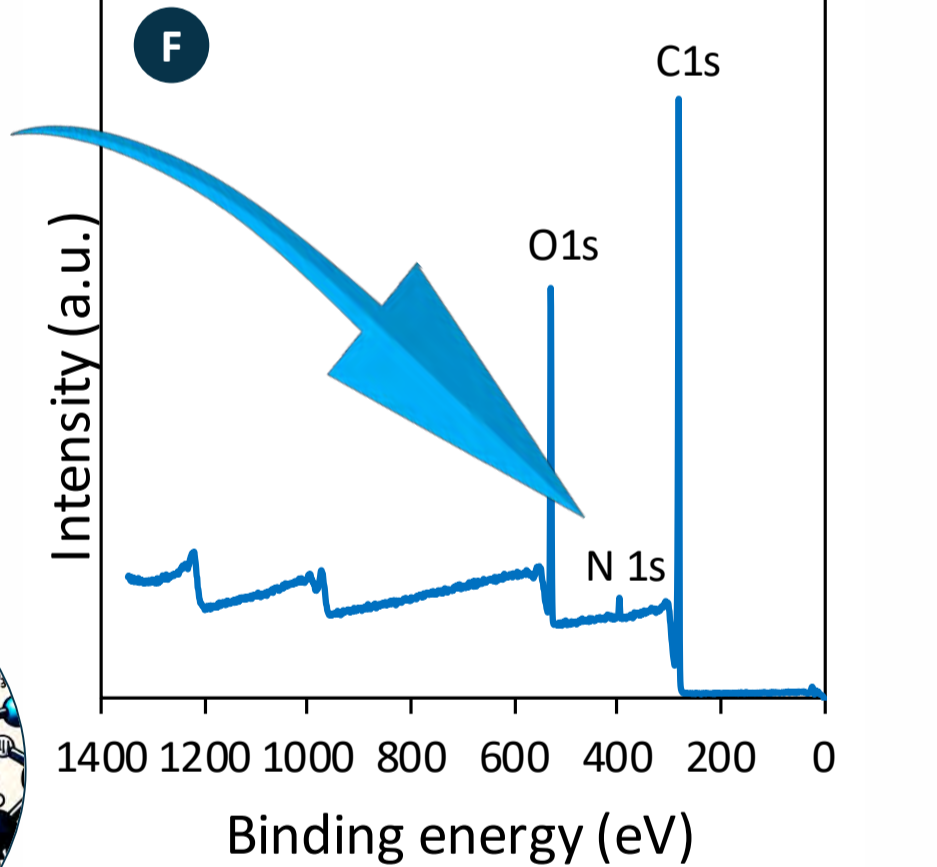
#### 4.2 FURTHER OPTIMIZATION

##### Laser modification

The application of a laser beam to the catalytic electrode with potato peels hydrochar **D** promoted: 1) the hydrochar melting into supported nickel foam electrode, favouring electronic transference and stability 2) the generation of oxygenated groups, favouring OER. The laser beam intensity resulted to influence the OER performance.



Thiourea addition **E** allowed the introduction of nitrogen-based functional groups **F**, favouring OER performance. However, too high thiourea concentrations caused a detriment on OER performance due to pores clogging.

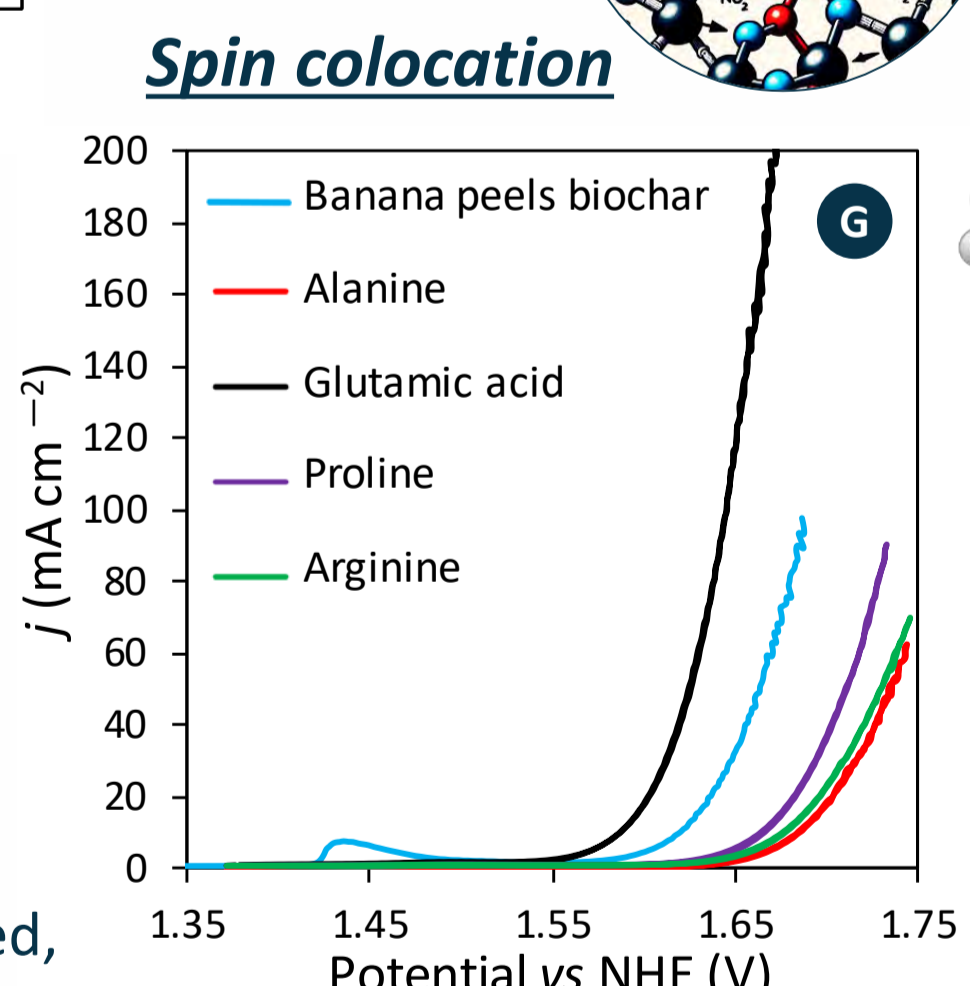


#### 4.3 STABILITY EVALUATION

To compete with metal-based catalyst there is a need of materials with not only good electrical behaviour but also long stability, thus it was evaluated **H**.

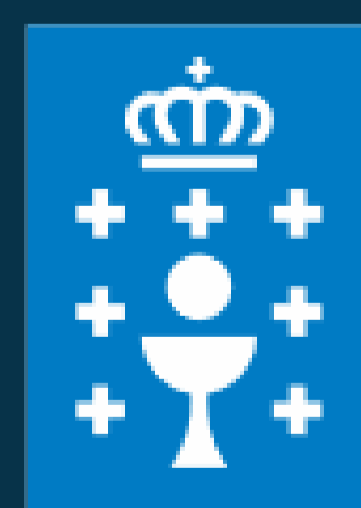
<b>H</b> Voltage increase (%) for 100 mA	Potato peels hydrochar	Potato peels hydrochar laser 3,713 mJ/cm <sup>2</sup>	Banana peels biochar	Banana peels biochar + 5% thiourea	Banana peels biochar + 5% glutamic acid
Accelerated Degradation Test (8,000 CVs)	7	2	6	-6	-2
Chronometry (100 h)	8	6	11	9	6

The synthesized chars were even more stable after the studied modifications. Indeed, some of them were even activated during consecutive energy input (negative values).

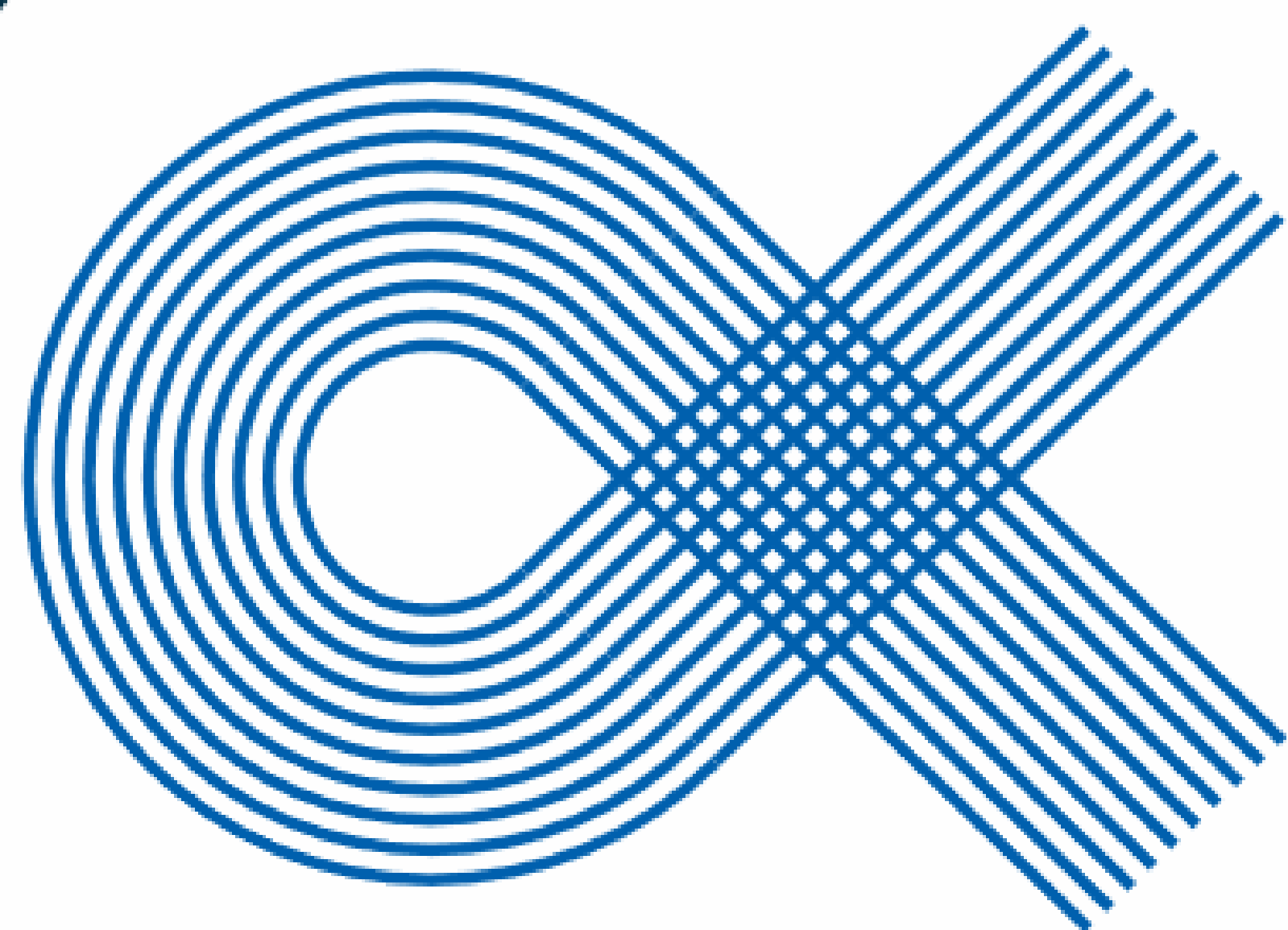


Chiral compounds addition favour spin collocation on oxygen atoms, favouring OER **G**. Thus, glutamic acid addition caused an extremely intensity increase, defeating metal-based catalysts.

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